

Lecture 3 - Atoms + Solids

Last time, introduced spin-statistics theorem:

integer spin = boson = + exchange
 half-integer spin = fermion = - exchange

Consequences:

- Two fermions can't occupy same single particle state (including spin)
- Two bosons can.
- Two fermions in different states more likely to be found far apart
- Two bosons more likely to be found close together
- Symmetry doesn't matter if single particle states have no overlap.

See
 supplement
 online

Sometimes think about this using "exchange forces"

"Quantum" force that attracts bosons + repels fermions

Force doesn't come from Hamiltonian, but it is real

→ takes a lot of energy to push two fermions together

Spin effects:

With spin, typically have

$$\Psi(\vec{r}_1, \vec{s}_1; \vec{r}_2, \vec{s}_2) = \Psi(r_1, r_2) \chi(s_1, s_2) \quad \text{- spin decoupled}$$

$$P\Psi(\vec{r}_1, \vec{s}_1; \vec{r}_2, \vec{s}_2) = \Psi(\vec{r}_2, \vec{s}_2; \vec{r}_1, \vec{s}_1)$$

Need overall ψ to have correct exchange

singlet:
 $X = \frac{1}{\sqrt{2}}(\uparrow\downarrow - \downarrow\uparrow)$
For instance, if $\chi(s_1, s_2) = \text{spin singlet}$
then $\chi(s_2, s_1) = -\chi(s_1, s_2)$

total $S=0$
Then for fermions, need $\psi(r_1, r_2)$ symmetric
bosons need antisymmetric

Last point:

How do we symmetrize wave function for more than two particles?

Say N states $\psi_a, \psi_b, \dots, \psi_n$
 N particles r_1, r_2, \dots, r_N

For bosons, just sum all permutations of states
particles:

$$\psi(r_1, r_2, \dots, r_N) = A \left[\psi_a(r_1) \psi_b(r_2) \dots \psi_n(r_N) \right. \\ \left. + \psi_b(r_1) \psi_a(r_2) \dots \psi_n(r_N) \right. \\ \left. + \dots \text{all other combinations} \right]$$

For fermions harder, need to keep track of signs. Can use "Slater determinant"

$$\psi(r_1, \dots, r_N) = \begin{vmatrix} \psi_a(r_1) & \psi_b(r_1) & \dots & \psi_n(r_1) \\ \psi_a(r_2) & \psi_b(r_2) & \dots & \psi_n(r_2) \\ \vdots & \vdots & \ddots & \vdots \\ \psi_a(r_N) & \psi_b(r_N) & \dots & \psi_n(r_N) \end{vmatrix}$$

Evaluate with normal rules for determinants

Can use Pauli exclusion principle to understand multi-electron atoms.

For instance, He has 2 electrons.

To first approximation, say they don't interact.
Then each electron in a single particle hydrogen state

Lowest energy: both in $1s$ state
only possible for spin singlet

First excited state, one e^- in $2s$ state

Could now have spin singlet or triplet,
which has lower energy?

If spin singlet, spatial Ψ is symmetric
 \Rightarrow electrons like to be close

Spin triplet, spatial part is antisymmetric
 \Rightarrow electrons further apart.

But since electrons both negative charge,
cost EM energy to have close together

\Rightarrow spin triplet state is lower. - correct

Section 5.2 discusses in more detail,
explains spectroscopic notation

I'll let you read it,

Now talk about solids, particularly metals

Solid = bunch of atoms meshed together

Metal = each atom gives up a few electrons
free to drift through material

That's why metals conduct

Put some electrons in on one side of wire

Take some out on other

Complicated system:

Electrons interact with ionic cores & each other

Simplest model: say electrons are free
particles in a box

Turns out to work tolerably well!

Develop model: Free Electron Gas

$$\text{Box potential } V(x, y, z) = \begin{cases} 0 & 0 < x < l_x \\ & 0 < y < l_y \\ & 0 < z < l_z \\ \infty & \text{else} \end{cases}$$

(can write $U = U_x(x) + U_y(y) + U_z(z)$)
each 1D infinite square well

So wave function separates

$$\psi(x, y, z) = X(x) Y(y) Z(z)$$

We know square well solutions:

$$\psi(x) = \sqrt{\frac{2}{l_x}} \sin \frac{n_x \pi}{l_x} x$$

integer $n_x > 0$

So 3D wavefunctions are

$$\psi_{n_x n_y n_z}(x, y, z) = \sqrt{\frac{8}{V}} \sin\left(\frac{n_x \pi}{l_x} x\right) \sin\left(\frac{n_y \pi}{l_y} y\right) \sin\left(\frac{n_z \pi}{l_z} z\right)$$

$V = \text{volume } l_x l_y l_z$

with energies $E_{n_x n_y n_z} = \frac{\hbar^2 \pi^2}{2m} \left(\frac{n_x^2}{l_x^2} + \frac{n_y^2}{l_y^2} + \frac{n_z^2}{l_z^2} \right)$

$$= \frac{\hbar^2 k^2}{2m}$$

for $\vec{k} = \left(\frac{\pi n_x}{l_x}, \frac{\pi n_y}{l_y}, \frac{\pi n_z}{l_z} \right)$

But that's not the whole story...

even free electrons satisfy exchange requirement

Only two electrons in each state (n_x, n_y, n_z)
 [those two e^- 's are in spin singlet]

So if we have a lot of e^- 's, we'll occupy
 a lot of states.

→ What is total energy of electron gas?

Use geometrical construction

Allowed wave vectors \vec{k} form grid in "k-space"

